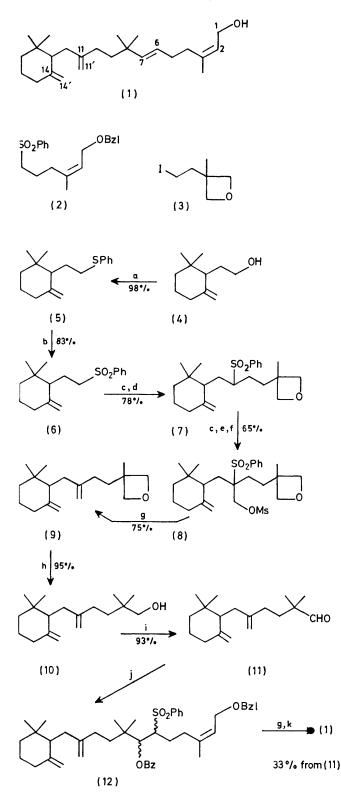
A Synthesis of Diumycinol

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The reductive-elimination of β -substituted sulphones was used to introduce the hindered alkene bonds at C(6)–C(7) and C(11) of diumycinol.

We report a new synthesis of diumycinol $(1)^1$ which is obtained by hydrolysis of the antibiotic diumycin.² The molecule was assembled from three principal fragments. The sulphone (2) was used previously in the synthesis of moenocinol³ and the oxetan (3) was prepared by the general method of Castro.⁴ The sulphone (6) was prepared in two steps from the known⁵ alcohol (4) by first converting (4) into the phenyl sulphide (5) using *N*-phenylmercaptosuccinimide and Bun_3P^6 followed by oxidation to the sulphone (6) using bis(trimethylsilyl) per-oxide^{7,8} in refluxing benzene as shown in Scheme 1.



Scheme 1. Bzl = benzyl, Ms = MeSO₂, Bz = benzoyl. Reagents and conditions: a, N-Phenylmercaptosuccinimide, Buⁿ₃P-CH₂Cl₂, 0 °C; b, 2.2 equiv. Me₃SiOOSiMe₃-benzene reflux, 18 h; c, BuⁿLi-tetrahydrofuran (THF), -78 °C; d, 1.5 equiv. of (3), -78 °C $\rightarrow 20$ °C; e, excess of CH₂O gas, -10 °C; f, MsCl-CH₂Cl₂, Et₃N; g, Na(Hg)-THF: MeOH (3:1), 0 °C; h, Li-H₂NCH₂CH₂-NH₂-Et₂O, 15 °C; i, (Me₂SCl)+ Cl⁻-CH₂Cl₂, -78 °C followed by Et₃N, -78 °C $\rightarrow 20$ °C; j, add (11) to lithium derivative of (2)-THF, -78 °C followed by BzCl, -78 °C $\rightarrow 20$ °C; k, Na-NH₃, -78 °C.

Alkylation of the lithium-derivative of the sulphone (6) with the oxetan (3) gave (7). However, a second alkylation of the lithium or potassium derivative of (7) with iodomethyltrimethylsilane under a variety of conditions failed and thereby foiled our use of the β -silylsulphone elimination⁹ which had served as a key step in the similar but sterically less demanding step in the moenocinol synthesis.³ Instead, the lithium derivative of (7) reacted readily with formaldehyde to give a produc which underwent methanesulphonation to give a diastereomeric mixture of β -methanesulphonyloxy-sulphones (8). The C(11) methylene group was then introduced by reductive elimination¹⁰ to give the diene (9). Reductive cleavage¹¹ of the oxetan (9) gave the alcohol (10) which was oxidized¹² to the aldehyde (11).

The synthesis was completed by condensing the lithium derivative of sulphone (2) with aldehyde (11). Benzoylation of the adduct gave a diastereomeric mixture of β -benzoyloxy-sulphones (12) which was treated with Na(Hg) to introduce the C(6)–C(7) double bond. Reductive debenzylation then gave diumycinol (1) whose 400 MHz ¹H n.m.r. spectrum (CDCl₃) gave well resolved ¹H olefinic resonances at δ 5.44 [t with further splitting, J 7.5 Hz, C(2)–H], 5.375 [br.d, J 16 Hz, C(7)–H], 5.26 [dt, J 16, J' 6.5 Hz, C(6)–H], and narrow multiplets at 4.725, 4.69 and 4.615, and 4.49 [C(14), H₂ and C(11), H₂ respectively]. There was no evidence for contamination of the product by the *cis*-isomer at C(6)–C(7).

There are three aspects of this synthesis which are noteworthy. Firstly, the reductive-elimination reactions used to introduce the olefinic bonds at C(6)-C(7) and C(11) were successful where alternative methods failed completely⁹ [in the case of the C(11) methylene] or failed to give the *trans*double bond at C(6)-C(7) selectively.¹³ Secondly, the oxidation of the sulphide (4) to the sulphone (5) by bis(trimethylsilyl) peroxide without affecting the double bond is potentially useful. A similar reaction on phenyl geranyl sulphide gave the corresponding sulphone in 83% yield. Finally, the use of an oxetan as an inert and sterically restricted latent neopentyl alcohol moiety could be useful in the synthesis of crowded complex molecules.

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References

- 1 P. A. Grieco, Y. Masaki, and D. Boxler, J. Org. Chem., 1975, 40, 2261.
- 2 W. A. Slusarchyk, J. A. Osband, and F. L. Weisenborn, Tetrahedron, 1973, 29, 1465.
- 3 P. J. Kocienski, J. Org. Chem., 1980, 45, 2037.
- 4 B. Castro, M. Ly, and C. Selve, *Tetrahedron Lett.*, 1973, 4455; B. Castro and C. Selve, *ibid.*, 1973, 4459.
- 5 S. Julia, M. Julia, H. Linarès, and J-C. Blondel, Bull. Soc. Chim. Fr., 1962, 1952.
- 6 K. A. M. Walker, Tetrahedron Lett., 1977, 4475.
- 7 P. G. Cookson, A. G. Davies, and N. Farzel, J. Organomet. Chem., 1975, 99, C-31.
- 8 Oxidation of simple sulphides to sulphones has been reported:
 D. Brandes and A. Blaschette, J. Organomet. Chem., 1974, 73, 217.
- 9 P. J. Kocienski, Tetrahedron Lett., 1979, 2649.
- 10 P. J. Kocienski, Chem. Ind. (London), 1981, 548, and references cited therein.
- 11 R. R. Savers, W. Schinski, M. M. Mason, E. O'Hara, and B. Byrne, J. Org. Chem., 1973, 38, 642.
- 12 A. J. Mancuso, S.-L. Huang, and D. Swern, J. Org. Chem., 1978, 43, 2480.
- 13 R. M. Coates and M. W. Johnson, J. Org. Chem., 1980, 45, 2685.